

GINZBURG, D.M.; MITKEVICH, N.D.

Theory of the commercial production of sodium hydrosulfide from
gaseous mixtures. Ukr.khim.zhur. 25 no.1:129-133 '59.

1. Khar'kovskiy nauchno-issledovatel'skiy institut osnovnoy khimii.
(Sodium sulfides) (MIRA 12:4)

5(4)

AUTHOR:

Ginzburg, D. M. (Khar'kov)

SOV/76-33-5-26/33

TITLE:

Heat Capacity and Integral Solution Heats in the System
NaOH - H₂O (Teployemkost' i integral'nyye teploty rastvorenija
v sisteme NaOH - H₂O)

PERIODICAL:

Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 5,
pp 1087 - 1092 (USSR)

ABSTRACT:

The author points out that the characteristic values mentioned in the title have hardly been investigated for high concentrations and high temperatures although these data are of importance in technical practice. C_p was measured in the concentration range of from 60 - 70 weight per cent NaOH and at temperatures of from 322 - 750°C. The data are shown in table 1. Hence it appears that the heat capacity of concentrated NaOH solutions is similar to that of NaOH melts. A formula for C_p is derived for high concentrations. On account of the comparable data on heat capacities of dissolved and melted NaOH it may be assumed that the structure of concentrated NaOH solutions is determined by the structure of the melt. The water molecules are distributed within this structure. The

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Heat Capacity and Integral Solution Heats in the System NaOH - H₂O SCV/76-33-5-20/33

integral solution heats of NaOH - H₂O were calculated for the range of from 50 - 350°C. Figure 1 shows the isothermal lines of the diagram enthalpy - concentration in the system NaOH - H₂O at 25°C, 93.33°C, and 322°C. The integral solution heats are shown in table 2, the temperature coefficients in figure 2. The temperature coefficient between 300 and 350° was not computed as there occur phase transitions in this temperature interval. The variations of the integral solution heats follow the laws for relatively diluted electrolytes at temperatures up to 75°C found by Yu. M. Kaganovich and K. P. Mishchenko (Ref 8). With a concentration increase of soda lye from 88.68 to 100 mol NaOH/ 1000 g the sign of the integral solution heats becomes positive. There are 2 figures, 2 tables, and 13 references, 8 of which are Soviet.

ASSOCIATION: Nauchno-issledovatel'skiy institut obozreniya khimii Khar'kov (Scientific Research Institute of Fundamental Chemistry, Khar'kov)

SUBMITTED: July 10, 1957

Card 2/2

5(1)
AUTHOR:

Ginzburg, D. M., Candidate of Chemical Sciences

06222

SOV/64-59-6-14/28

TITLE:

The Thermal Conductivity of Lime Obtained by Roasting Lime-stone at Different Temperatures

PERIODICAL:

Khimicheskaya promyshlennost', 1959, Nr 6, pp 510 - 513
(USSR)

ABSTRACT:

In a paper published in 1954 (Ref 1) it was pointed out that rocks of similar specific gravities coming from different deposits possess also similar coefficients of thermal conductivity. Limestone from the Golubovskoye deposit and chalk from the Golosnikovskoye and Raygorodskoye deposits as well as lime obtained by roasting Golosnikovskiy chalk were investigated. The thermal conductivity coefficients of lime obtained by roasting chalk, however, are different from those obtained by roasting limestone. In order to complement the data given in (Ref 1) the thermal conductivity coefficients of lime obtained from Gchikovskiy limestone were investigated in the present case. Lime was obtained at different roasting temperatures and had been left in the furnace at the same

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The Thermal Conductivity of Lime Obtained by Roasting SOV/64-59-6-14/28
Limestone at Different Temperatures

temperature for varying lengths of time after roasting. The thermal conductivity coefficient of lump lime was determined according to a method previously used (Ref 1), which had been developed by the fiziko-khimicheskaya laboratoriya Nauchno-issledovatel'skogo instituta ogneuporov (Laboratory of Physical Chemistry of the Scientific Research Institute of Refractories). A description of the apparatus is given in references 1 and 2. The lime samples were obtained by roasting (at 925, 1025, 1125, 1200, 1250, and 1350°) cylindrical samples produced by turning limestone. A laboratory shaft kiln (Fig 1) was used. The temperature was controlled by means of a platinum/platinum-rhodium thermocouple connected with the apparatus ERM-47. On the basis of the experimental values obtained equations for the calculation of the true thermal conductivity coefficients of the samples under investigation are given and the following generalized equation derived:

$$\lambda_t = -1.011 - 0.066 \cdot 10^{-2} t + 1.513 \cdot 10^{-3} \gamma, \text{ (where } \lambda_t = \text{true thermal conductivity coefficient of lump lime obtained from limestone at temperature } t, \gamma = \text{specific gravity of the lime).}$$

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APPROVED
The Thermal Conductivity of Lime Obtained by Roasting Limestone at Different Temperatures

06222

SOV/64-59-6-14/28

It is recommended to assume $\gamma = 1600 \text{ kg/m}^3$ (corresponding to a volume shrinkage of about 10%) in calculating the heat of lime production from limestone on the basis of the above equation, while formula $\gamma = 0.79 - 0.00049 t$ (Ref 1) should be used when lime is obtained from chalk. The thermal conductivity coefficients of lime obtained from limestone of four different specific gravities were calculated and the following tables given: Table 1, characteristic data of the roasting process, table 2, temperature dependence of the thermal conductivity of lime from Golubovka limestone, table 3, comparison of the results obtained for limestone samples from the above deposits, table 4, thermal conductivity coefficients of the Golubovskiy limestone at various temperatures, table 5, coefficients of the thermal conductivity of lump lime. There are 2 figures, 5 tables, and 6 references, 5 of which are Soviet.

ASSOCIATION:

Card 3/3

Nauchno-issledovatel'skiy institut osnovnoy khimii, NIOkhim
(Scientific Research Institute of Fundamental Chemistry,
NIOkhim)

GINZBURG, D.M. (Khar'kov)

Thermal properties of NaOH and H₂O in concentrated sodium hydroxide solutions. Zhur. fiz. khim. 36 no.4:747-751 Ap '62.
(MIRA 15:6)

1. Nauchno-issledovatel'skiy institut osnovnoy khimii.
(Sodium hydroxide--Thermal properties)

JOHNSON, H.M.; CALIFORNIA, U.S.A.; LITHIUM, 99%

Density of potassium carbonate solution. 100% Purity
37 No. 112-353-2327 N 16d (100% Purity)

1. Muriatic acid added to 100% potassium carbonate solution.

SCHERF, L.A.; HILLMAN, H.S.; FISHER, V.J.

Electrolytic reduction of carbonate solutions. *J. Am. Chem. Soc.* 39
no. 1, p. 92, 1917. P. 164. (Chem. 18:3)

L. I. Ivanov-Dobrovolskiy Institut gornojekhimi, Rastkoy.

GINZBURG, D.M.

Density of soda-potash solutions and the vapor pressure over them. Zhur. prikl. khim. 38 no.1:55-58 Ja '65.

(MIRA 18:3)

1. Nauchno-issledovatel'skiy institut osnovnoy khimii, Khar'kova.

GINZBURG, D.M.; PIKULINA, N.S.; LITVIN, V.P.

Sistem NH₃ - P₂S - H₂O. Zhur. prikl. khim. 38 no.9 2117-2119
S '65. (VIZ, '65-11)

1. Nauchno-issledovatel'skiy institut otsenivayushchii,
Khar'kov.

GINZBURG, D.M.

System NH₃ - CO₂ - H₂O. Zhur. prikl. khim. 38 no.10:2197-2210
0 '65. (MIR 18:12)

1. Nauchno-issledovatel'skiy institut osnovnoy khimii, Khar'kov.
Submitted May 27, 1963.

GINZBURG, D.S.

Erythema nodosa as a trichophytid. Vest.derm. i ven. 32 no.5:73
S-0 '58 (MIRA 11:11)

1. Iz Vologodskogo oblastnogo vendispansera.
(ERYTHEMA)

GINZBURG, D.S.

Clinical and histological aspects of Darier's disease. Vest. derm.
i ven. 33 no.2:25-28 Mr-Ap '59. (MIRA 12:7)

1. Iz Vologodskogo oblastnogo vendispansera (glavnnyy vrach Ye.K.
Savashkevich).

(KERATOSIS POLLICULARIS, case reports,
(Rus))

GINZBURG, D. S.

Trichomycoses and their pathogens in Vologda Province during
1949-1958. Vest. derm. i ven. no.3:71-74 '62.
(MIRA 15:6)

1. Iz Vologodskogo oblastnogo kozhno-venerologicheskogo
dispansera (glavnnyy vrach Ye. K. Savashkevich)

(VOLOGDA PROVINCE—HAIR—DISEASES)

~~REINHOLD~~, D.A.

Utilization of peat in the socialist agriculture. Czechoslovakian agricultural review 1957, 17 p.

PROCESSES AND PROPERTIES INDEX

Adsorption activity of peat and coals. N. P. Karpolevich and D. Z. Gindus. *Bol'shov Akad. Nauk, Inst. Khim. Sverdlov. Polz.*, 1, 143 (4) (1934).—The adsorption activity of peat, brown and anthracite coals, and coke of 0.25-0.5 mm. was detd. with solns. of I⁻ in KI and with oxalic acid. The sample (1 g.) was shaken in a closed 200-cc. vessel contg. 80 cc. of the soln., allowed to stand for 1 hr. and 25 min., shaken again and filtered. The first portions of filtrate were discarded, but the remainder was used for titration. Adsorption follows $x/m = KC^n$. A decrease in the following order: peat; brown coal; anthracite. Differences between actual and theoretical results fall within the exptl. errors. D. Z. Kamich

21

ASD-SEA METALLURGICAL LITERATURE CLASSIFICATION

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Relation between adsorption, solubility and solvent polar properties. N. P. Ermolenko and D. Z. Gamburg. *Zh. (U. S. S. R.)* 3, 831-83 (1937). —With anthranilic acid (I) the relations are quite complex. In solvents composed of 2 nonpolar components ($\text{CHCl}_3\text{-CCl}_4$) or of 1 polar and 1 nonpolar component (PhMe-CCl_4) the adsorption (A) and solv. (L) of I vary inversely.

In a pair of solvents like $\text{CHCl}_3\text{-PhMe}$, similar in structure

and close in the values of their polar constants, the changes in A and L are nearly parallel. The adsorption of charcoal of I from mixed solvents composed of a polar and nonpolar component, the latter affecting the polarity of the first ($\text{H}_2\text{O-H}_2\text{C}_6\text{H}_5$), passes through a min. L in such cases increases with increase in the amt. of polar component in the mixt. In a mixt. of 2 strongly polar solvents like $\text{KOH-H}_2\text{O}$ and $\text{Me}_2\text{CO-KOH}$ A and L vary inversely.

John Livak

G/NZ BHP G P

PROCESS AND PROPERTIES

Ca

Importance of the constitution of organic acids for their adsorption from mixed media. N. F. Ermolenko and D. Z. Gerasimov. *Colloid J.* (U. S. S. R.) 5, 263-70 (1937); cf. *C. A.* 33, 6112^c.—Adsorption by wood charcoal from aqueous solvents—increases in the series AcOH , chloroacetic, dichloroacetic, trichloroacetic and salicylic acid. For a given acid it increases from RIOH to $\text{C}_6\text{H}_5\text{OH}$ and CCl_4 . The adsorption from mixed solvents $\text{RIOH} + \text{C}_6\text{H}_5\text{OH}$ shows a min., whereas there is neither max. nor min. in the curve of adsorption from the mixta. $\text{CCl}_4 + \text{C}_6\text{H}_5\text{OH}$ + H_2O . Salicylic acid was also adsorbed from $\text{C}_6\text{H}_5\text{OH}$ + $\text{C}_6\text{H}_5\text{COOH}$ (a min.), and α -aminobenzoic acid from H_2O + $\text{C}_6\text{H}_5\text{COOH}$ mixts. L. I. Bakeman

430.51 METALLURGICAL LITERATURE CLASSIFICATION

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Swelling of vulcanized rubber in mixed media. N. F. Ermolenko and D. Z. Ginzburg (Acad. Sci. Byelorussian S.S.R., Minsk). *Kolloid Zhur.* **19**, 182-7 (1951). A vulcanized rubber made at 143° from natural rubber (100), S.I. 30, condensate K-11 1.25, addol 1, ZnO 25, C black 3%, chlk 30, stearic acid 1, and pine tar 2 parts, was up, at 18°, to vol. of CCl_4 , 2.7 CaH_2 , 4.5 CHCl_3 , 0.010 Ba(OH)_2 , or 1.0 vol. MeCO . The dependence of the vol. taken up on the composition of binary mixtures of these solvents was found only for $\text{CCl}_4\text{-CaH}_2$. In the other mixts., the max. swelling was greater the smaller the mol. polarization of the mixt. The rate of attainment of equil. was great in CaH_2 , smaller in CCl_4 , and smaller still in CHCl_3 and MeCO .

G INEL-GAUGER DRIVAW

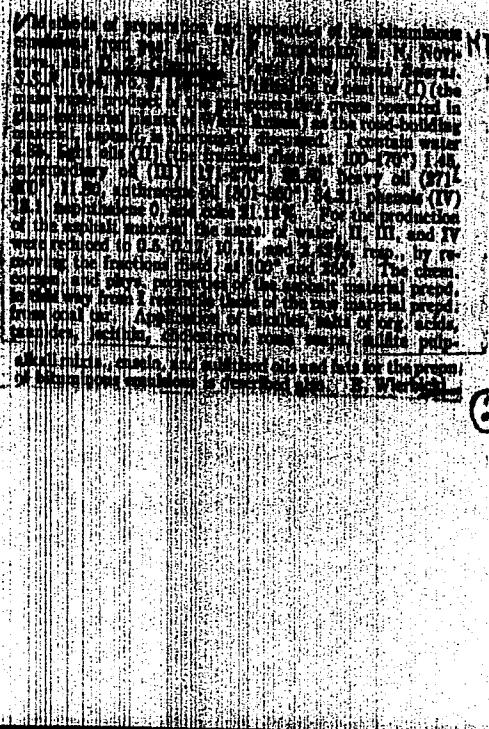
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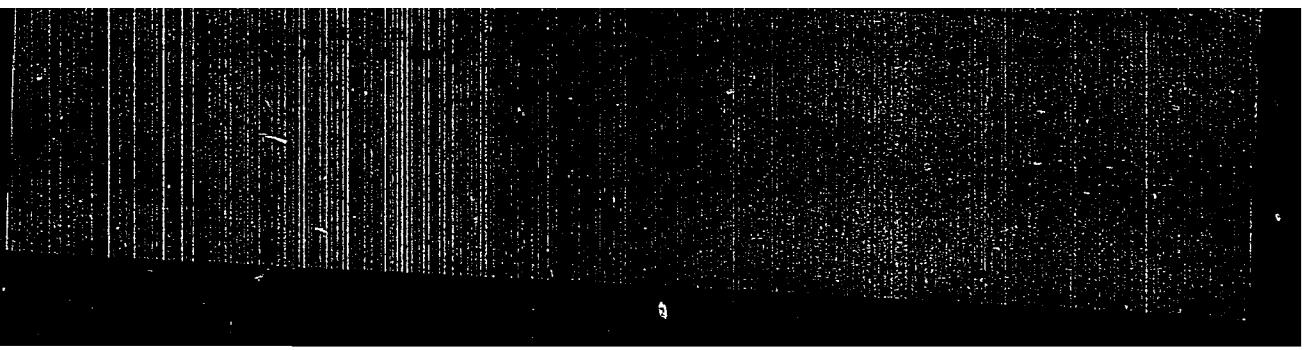
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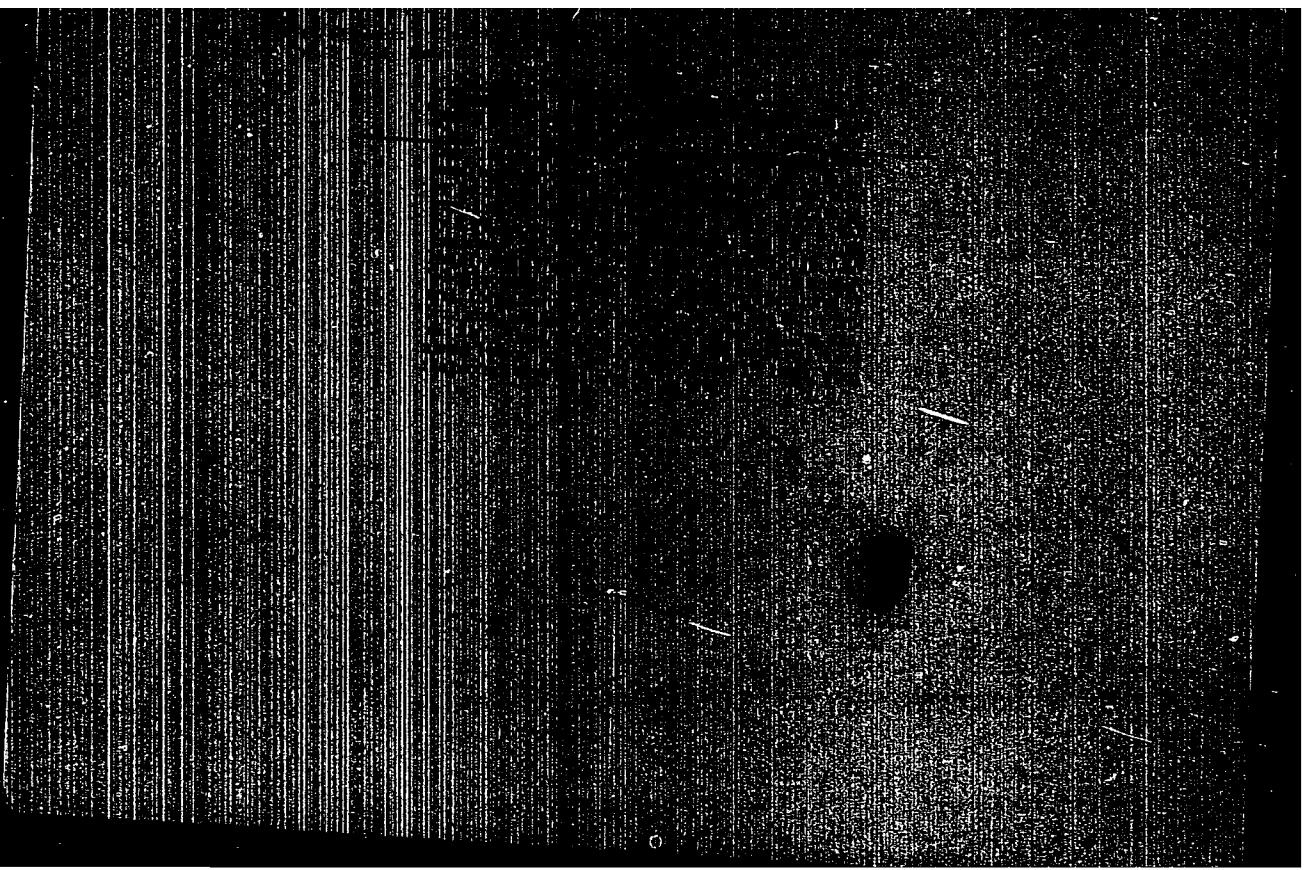
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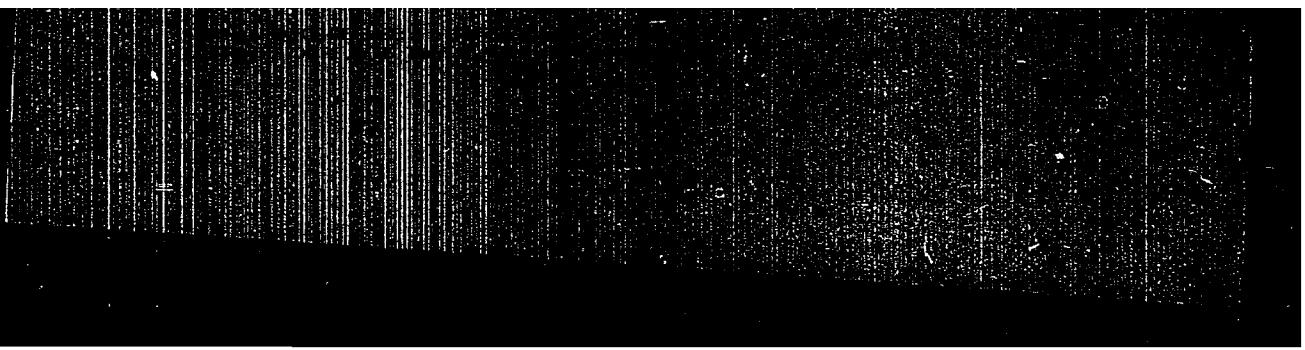
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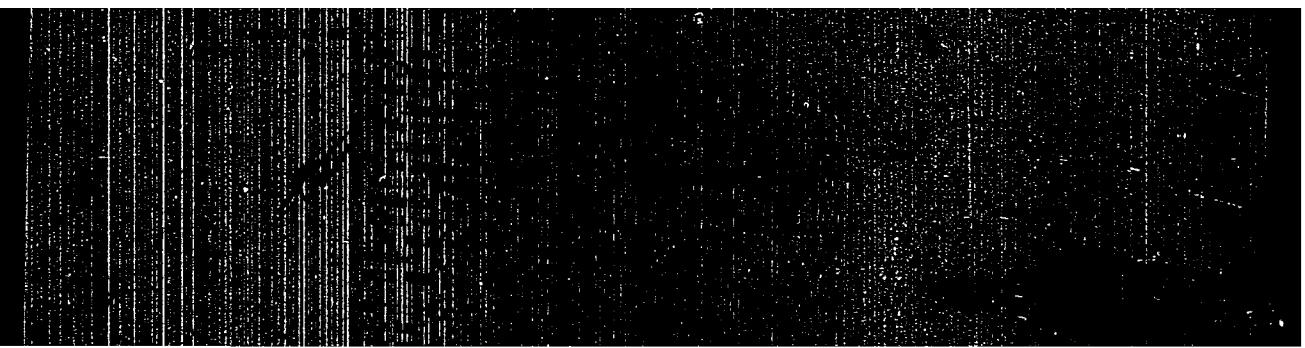
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GINZBURG, D. Z., and YERMOLENKO, N. P.

"Change of the Symmetry of Molecules of Vegetable Alkaloids Under the Influence of Denaturizing Substances" (Изменение симметрии молекул алкалоидов под влиянием денатурирующих веществ) from the Book Trudy of the All-Union Conference on Colloid Chemistry, pp. 397-403, Iz. AN BSSR, Minsk, 1959

(Report given at above Conference, Minsk, 21-24 Dec. 1958)

Yermolenko: Act. Mor. RA BSSR

GINZBURG, D.Z.

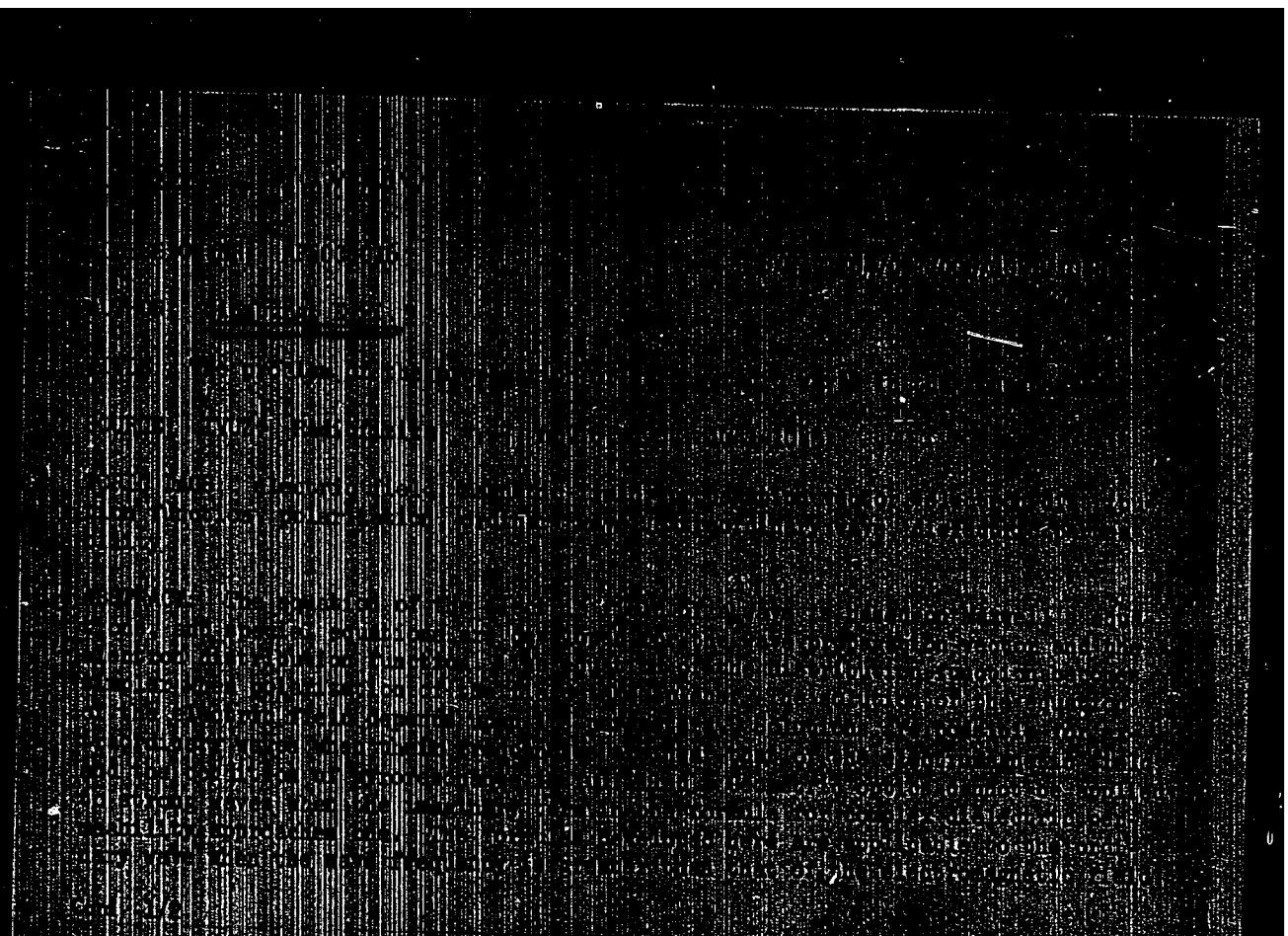
Relation between the structure of chemical compounds and their
toxic effect on the zoosporangia of *Synchytrium endobioticum* (Schilb.)
Perc. Vestsj AN BSSR. Ser. biol. nav. no.3:31-38 '60.

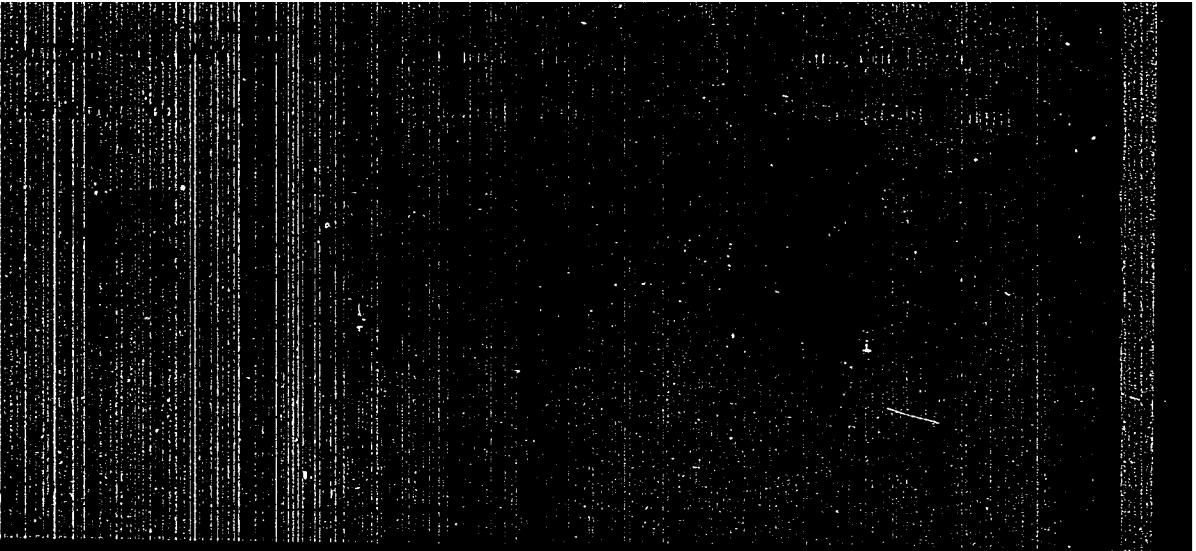
(WHITE RUSSIA--POTATO WART) (MIRA 14:1)
(BENZENE) (FUNGICIDES)

MANDAL, B.N., VASUDEV, I.S., PRAKASH, M. (1961)

"The Role of Protein-Lipid Complexes and Lipid-protein Particles in the Early Stages of Lipid Structure (A Contribution to the Theory of Lipidosis and the Pathogenesis of Alcohol.)"

Received from the Bio-Physics Division, Central Research Institute, TIFAC and I.I.T.,





LYAGIN, I.V.; GINZBURG, E.Kh.

Σ^+ $\rightarrow p + e^+ + e^-$ and $\Sigma^- \rightarrow p + \mu^+ + \mu^-$ decays. Zhur.eksp.i
teor.fiz. 41 no.3:915-918 S '61. (MIRA 14:10)

1. Smolenskiy gosudarstvennyy pedagogicheskiy institut.
(Particles (Nuclear physics))

IVANOV, E.A.; VIDINEYEV, L.P.; GINZBURG, E.L.; MAZUK, V.B.

Tectonic development of the lower Paleozoic of the southern
part of the Siberian Platform. Neftegaz. geol. i geofiz. no.
10:12-15 '64 (MIRA 18:1)

1. Gosudarstvennyy trest po geologicheskim izyskaniyam na neft'
v Vostochnoy Sibiri.

GINZBURG, E.L., SAMSONOV, V.V.; FUKS, R.A.

Prospecting gas fields in the Irkutsk oblast (part). Neftegaz.
geol. i geofiz. no.10(24-25) '64 (USSR 18:1)

1. Gosudarstvennyy test. po neftyanym i gazonym neftam
v Yugo-zhnoy Sibiri.

GINZBURG, E.L.

Krivolutskoye swell-shaped uplift is a new zone of possible oil
and gas accumulations. Neftegaz. geol. i geofiz. no. 10:3-6
'65. (MIRA 18:12)

1. Trest "Vostsibneftegeologiya".

3906. **Malignant pancreatic cysts and their X-ray diagnosis** (Russian text) BRAITSEVA N. N. and GINZBURG E. M. *Klin. Med. (Mosk.)* 1958, 36/4 (81-85)

A malignant tumour is found in 9% of benign pancreatic cysts. According to Kennard, about 25 cases of malignant pancreatic cysts have been described up to 1940; 2 reports on this type of tumour were found in the Russian literature (1935 and 1938). The authors give a detailed description of 2 additional cases. In both, the clinical picture, and especially an exact X-ray diagnosis, are presented. Both cysts were examined microscopically and recognized as malignant. A survey of the X-ray symptomatology of pancreatic cysts establishes that the diagnosis of malignant pancreatic cysts is only possible through close cooperation between the clinical physician and the X-ray diagnostician. The greatest attention must be paid to the natural history of the disease, which can give indications as to the malignancy of the process.

Brückner - Ostrava

GINZBURG, E.M.

Radiography of the temporal bone in a Stenvers projection.
Vest.otorin. 21 no.4:84 J1-Aug '59. (MIRA 12:10)

1. Iz rentgenologicheskogo otdeleniya (zav. E.M.Ginzburg)
Moskovskoy gorodskoy bol'nitsy No.58.
(TEMPORAL BONE--RADIOGRAPHY)

GINZBURG E.M. (Moskva, Ye-24, 3-ya Kabel'naya, 3. kv.131)

Hemangioma of the diaphysis of the humerus. Vop onk. 8 no. 10:
84-85 '62. (MIRA 17,7)

1. Iz rentgenologicheskogo otchaleniya (zav. - E.M.Ginzburg)
Moskovskoy gorodskoy bol'nitsy No. 58 (glavnnyy vrach - dotsent
Ye.Ya.Khesin).

GINZBURG, E.M.

Five years experience with the use of an obturator-adapter.
Vrach.delo no.13148-149 Ja '63. (MIRA 16:2)

1. Rentgenologicheskoye otdeleniye (zav. - E.M. Ginsburg) Moskovskoy gorodskoy bol'nitsy No. 58.
(ENEMA)

GINZBURG, E.M.

Excessive development of the mucous membrane of the stomach.
Vestn. rentgen. i radiol. 38 no.4:71-73 Jl-Ag'63
(MIRA 17:2)

1. Iz rentgenovskogo otdeleniya (zav. E.M.Ginzburg) Moskov-
skoy gorodskoy klinicheskoy bol'nitsy No.58 (glavnyy vrach-
dotsent Ye.Ya. Khesin).

TVEROVNIK, M.S., p. 22 K.BUDZ, R.M.

Osteoplastic hypernephroid cancer of the kidney. Urologist.
29 no.3 49-53 My. 12. 64. (L76-12-10)

Остеопластична гипернеприодна рака със злокачествени кости. Урологист.
1964 година № 3, стр. 49-53. Тверовник, М.С. Буздз, Р.М.

GINZBURG, E.N., red.

[Processing of liquid media; papers of the Scientific Research Institute for Fertilizers and Insectifuges] Obrazotka zhidkikh sred; trudy NIUF. Kiev, Laboratoriia nauchno-tekhn. informatsii, 1962. 62 p. (MIRA 17:4)

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GINZBURG, E.N., kand.tekhn.nauk

Comparative technological characteristics of band and trough
vacuum filters. Khim.mashinostr. no.2:9-12 Mr.Ap '64.

(MIRA 17:4)

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GRIMES, Paul, (Paul E. Grimes)

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PLATE I BOOK INFORMATION 30V/4A/65

EXPERIMENTAL AND MATERIAL PROBLEMS IN METALS
(Corrosion, Fracture, Creep, Plastic Deformation, Strength, Etc.)

Moscow, Sovzavizdat, 1957, 215 p., 2,500 copies printed.

Ed. (Title Page): I.M. Lachinov; Candidate of Technical Sciences, Ed. (Table of Contents):

I.M. Lachinov; Trans. Ed.: V.I. Borodov.

ABSTRACT: This collection of articles is intended for technical personnel of power

stations, power ministry plants, and scientific research institutes.

EDITORIAL NOTE: The articles set forth the results of investigations that were conducted

by Order Ministry Aviation, Aerospace, Telecommunications Institute (Lavrent'ev, P.S.
Sternovskiy) (Department of Metals of the All-Union Inst. for Testing Materials

Lavrent'ev, P.S. Directorship) in the years 1953-57. The articles deal with the problem

of investigating new types of steel and of analyzing the causes of damage or destruc-

tive parts of power plant equipment. Problems associated with operating depend-

ability of nuclear plants in view of piping for high and extreme-pressure boilers or turbines

under such temperature conditions are given. The reasons for failures and methods

together with remedial joints and methods for remanufacturing damaged equipment

discussed. No generalities are mentioned. References accompany individual

articles. D.M. and R.Y. Masell. On British Practice, Structure and

Properties of Welded Joints of High Pressure Steel Pipe

47

Khromov, A.A. and V.I. Sviridov. The Effect of the Complex State of

Tension and Stress on Mechanical Properties of Pipes

52

Khromov, A.A. Investigation of Properties of Commercial-Type Iron and

Steel

57

Klimov, I.M. and I.I. Podgornyy. On the Effects of Temperature Changes

on the Creep Strength of Low-Carbon Steel

63

Klimov, D.M. and V.I. Sviridov. Dependence of the Creep Deformation of

Stainless Chrome Steel on its Structure and Mechanical Properties

69

Klimov, V.I. and I.M. Lachinov. The Ability of Metals for Dry Pressurized

Boiling

Klimov, V.I. and I.M. Lachinov. Oxidation-Passivation of Corroded

Mechanical Parts

Klimov, V.I. Protection of Thermocouple Joints from Corrosion and Boiling

206

Klimov, V.I. Protection of Thermocouple Joints from Corrosion and Boiling

213

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CONTENTS.—This collection of 15 articles deals with various problems in the production of better wheat. Special attention is paid to the mechanism of deformation of wheat grains, oxygen, iron, and water. Various methods and features of seeds are analyzed, and means for increasing the heat, resistance and plasticity are described. During the sprouting, the heat, resistance and elasticity conductivity of the germinating wheat are studied, and the mobility of water in the grain is determined. A method of determining the water content of the grain is proposed. The influence of the temperature of the grain on its resistance to insects is investigated.

the American Society for the Advancement of Science, Boston, Mass., April 10-14, 1900.

BRITISH, IRISH, AND AMERICAN *Geological Society* **MEMBERSHIP** **REGISTRATION** **LISTS**

Bernard L. E. and Dr. D. G. Denton. Possibility of Differentiating the
Benzylidene Derivatives According to the Ratio 4/2 and 4/3/2
From the Angle of the Possibilities of Synthetic Lines.

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REVIEW. *Eric Hobsbawm on Change in Industrial Forces: Industrialisation, Structure or Reform? From Marx to Marxists*, by Eric Hobsbawm. London: Tavistock, 1970.

Kiln of a Japanese in Steel with 15% Cr and 14% Ni
Early, No. 22— and Dr. K. H. Finkler—*Stainless Structural Stabilization
and Its Effects on Decreasing the Durable Hardening and Plasticity
of Alloys*

ATLANTIC, 2 M., 2 M. Valdina, and 2 M. Puntalito. Investigation of the behavior and the place of a difficult situation during a prison

REVIEW ARTICLE The Nature of the "White Group" in Wells of the
Neogene Pacific Slope
By V. F. and E. L. BROWN, Geological Survey of Canada, Ottawa, Ontario, and A. H. DAWSON,
University of British Columbia, Vancouver, B.C.

REVIEW.— L. E. DURR and E. F. BROWN. Effect of the Concentration of the Oxide of Hydrogen on the Growth Curves of Allium.

Symposium, Vol. 6, 2012. **Editor-in-Chief**, and **Guest Editors**: **Presentations in Chemical Physics: Theory and Challenge by the Means of Transformational and Real Analysis**

STRUCTURE OF Iodo-, and Vito, PHOSPHATE, INVESTIGATION OF THE NATURE OF
CATIONIC CRYSTALLIZATION

S/127/62/000/004/097/201
A052/A101

E. S.
AUTHORS: Vidman, D. N., Ginzburg, E. S.

TITLE: The dependence of the damping decrement of stainless chromium steel
on the structure state and mechanical properties

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 4, 1962, 26, abstract 41152
(V sb. "Ekspluatats. nadezhnost' metallicheskikh ustroystv".
Moscow-Leningrad, Gosenergoizdat, 1959, 8)-07)

TEXT: The damping decrement was determined by the data of measurement of
the amplitude of free oscillations at bending the steam turbine blades. Mechani-
cal oscillations imparted to a cantilever-fastened blade were then transformed
in electric ones by means of an induction pickup and recorded with a loop
oscilloscope. Maximum bending stress at the root of the working part was 350
kg/cm². More than 200 blades made of material corresponding by the chemical
composition to 1X13 (1Kh13) and 2X13 (2Kh13) grades of steel were investigated.
Out of the mentioned blades samples were made which were subjected to mechanical
tests. Furthermore, an investigation of microstructure of blades with different
damping decrement values was carried out. The presence in the structure of

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The dependence of the damping ...

S/17/62/000/004/097/201
A052/A101

excess phases, - free chromium ferrite or free Cr carbides at the boundaries of grains, - reduces the damping decrement. A perlite structure without excess phases secures maximum damping decrement value; δ_f , δ_s , hardness and δ for blades with different damping decrements practically do not change; a_k and ψ increase with an increase of the damping decrement. An increase of the damping decrement by a factor of 1.5 - 2 (from $\rho = 0.0171$ to $\rho = 0.0301$) has just a little effect on δ_w . The obtained data permitted the recommendation of the following optimum composition of stainless Cr steel for working turbine blades: 0.15 - 0.20% C, 11.5 - 13.0% Cr, 0.6 - 0.8% Ni. There are 7 references.

M. Matveyeva

[Abstracter's note: Complete translation]

Card 2/2

81812

S/096/60/000/08/012/024
E194/E484

26.1000

AUTHORS: Lipshteyn, R.A., Khaykina, S.E. and Ginzburg, E.S.,
Candidates of Technical Sciences

TITLE: The Resistance of Gas Turbine Metals to Vanadium²⁷
Corrosion²³

PERIODICAL: Teploenergetika, 1960, Nr 8, pp 57-60 (USSR)

ABSTRACT: The use of sulphurous fuel oil in gas turbines is associated with vanadium corrosion of the blades at temperatures above 625°C. Vanadium is present in the fuel oil in the form of metallo-organic compounds and sodium in the form of sodium chloride. During the process of combustion the vanadium oxidizes to V₂O₅ and the sodium chloride is converted into sodium sulphate. Tests were made in which samples of steel, 6 mm diameter and 30 mm long, were immersed to a third of their height in ash of known composition. Samples that had been treated in this way were placed in an electric furnace where the temperature was maintained constant for periods up to 60 hours with a steady flow of air. After cooling, corrosion products were removed from the samples, either mechanically or by chemical means. Tests were made on

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E194/E484

The Resistance of Gas Turbine Metals to Vanadium Corrosion

austenitic chrome nickel steel grades EYalT^R, EI-405,¹²
EI-612,¹³ and nickel-base alloys of the Nimonic type,
see Table 1. Test results with ash containing various
amounts of V₂O₅ and Na₂SO₄ when corroding steel grade
EYalT at temperatures of 625, 750 and 800°C, are given
in Fig 1. At all temperatures there is a clear maximum
in the corrosion corresponding to an ash containing
87% V₂O₅ and 13% Na₂SO₄. Pure vanadium pentoxide
causes relatively little corrosion at temperatures
below 750°C and pure sodium sulphate causes relatively
little corrosion at temperatures up to 800°C. The
composition of the most corrosive mixture corresponds to
a compound of formula Na₂O · V₂O₄ · 5V₂O₅ which has a
melting point of 625°C. It is of interest to compare
the corrosion of steel EYalT with this artificial
mixture of vanadium pentoxide and sodium sulphate with
corrosion obtained under practical conditions. Data on
corrosion of this steel under practical conditions lies
surprisingly close to the corrosion curves with the
artificial ash at 750°C, see Fig 1. Tests with the ✓

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The Resistance of Gas Turbine Metals to Vanadium Corrosion

various steels were made within the temperature range of 625 to 800°C and durations of 15 to 60 hours with the most corrosive mixture of artificial ash. As will be seen from the results given in Fig 2, the temperature is a decisive factor and the rate of corrosion greatly increases with the temperature. Fig 3 shows the amount of corrosion products formed also increased with time; there is often an initial induction period followed by an auto-catalytic type of curve. The different grades of steel do not all perform in the same way at different temperatures and the differences are discussed. The corrosion products of different steels also differ in appearance. The low corrosion resistance of steel EI-405 is attributed to its 2.5% content of molybdenum. It is supposed that the molybdenum oxide MoO₃ formed during vanadium corrosion of the steel has a high vapour pressure at a temperature of 750 to 800°C which tends to throw the scale off the metal and to bare the metal surface to further corrosion. It is concluded that the use of molybdenum should be avoided in steels subject to vanadium corrosion.

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The Resistance of Gas Turbine Metals to Vanadium Corrosion

The nickel-base Nimonic alloy behaves better than chrome-nickel austenitic steel but it could not be successfully used in gas turbines burning high sulphur fuel oils at temperatures of 650°C and above since in the presence of the corrosive mixture of vanadium oxide and sodium sulphate, Nimonic alloy has a 12% loss of weight after 60 hours at 750°C and 18% at 800°C. The problem accordingly arose of improving the vanadium corrosion resistance of gas turbine blades of steels EI-405 and EI-612 by chemical-thermal treatment of the surface, saturating them with chromium, aluminium or nitrogen. To this end, samples of these steels were appropriately treated and the corresponding test results are given in Table 2. Treatment of steel EI-612 with chromium plus nitriding gives a considerable improvement in corrosion resistance at 750°C but increasing the temperature to 800°C completely removes this effect and even impairs the resistance of the steel to vanadium corrosion. Additional special investigations are required to elucidate the reason for this effect. It is

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E194/E484

The Resistance of Gas Turbine Metals to Vanadium Corrosion
interesting that platinum¹¹ porcelain and quartz are
also subject to vanadium corrosion at high temperatures.
There are 3 figures, 2 tables and 8 references
7 of which are Soviet (4 of these being Russian
translations from Proceedings of World Petroleum Congress)
and 1 English.

ASSOCIATION: Vsesoyuznyy teplotekhnicheskiv institut
(All-Union Thermo-Technical Institute)

ard 5/5

X

LAGUNTSOV, I.N., kand.tekhn.nauk; GINZBURG, E.S., kand.tekhn.nauk

Metal for principal models of new power equipment. Teploenergetika
7 no.5:3-12 My '60. (MIRA 13:8)

1. Vsesoyuznyy teplotekhnicheskiy institut.
(Steel) (Power engineering--Equipment and supplies)

34397
S/695/61/000/000/003/005
B139/B104

18.115/

AUTHOR: Ginzburg, E. S.

TITLE: Metal for turbines with high and supercritical steam parameters

SOURCE: Gorshkov, A. S., V. Ye. Doroshchuk, and N. V. Kuznetsov, eds
Povysheniye parametrov para i moshchnosti agregatov v
teploenergetike; sbornik statey. Moscow, Gosenergoizdat,
1961, 112 - 121

TEXT: The housings of the shutoff valves and steam chambers of the
CBK-150-1 (SVK-150-1) turbine for 150 Mw, 3000 rpm, 170 at and 550/520°C
were originally made from the high-alloy austenitic steel of type 1A-1
(LA-1). The manufacture of large castings from this steel is difficult
For this reason, welding together of individual forgings from ЭИ-405
(EI-405) steel was introduced at the LMZ Plant in cooperation with the
TsNIIITMASH and TsKTI. The SVK-150-1 turbine has austenite valve housings
in an outer perlite cylinder, and inner perlite cylinders in austenite jet
chambers. The ПБК-150 (PVK-150) 150-Mw turbine from the KHTGZ Plant and
the ПБК-200 (PVK-200) 200-Mw turbine from the LMZ Plant for the steam
Card 1/0 ✓

Metal for turbines with high and ...

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B139/B104

parameters of 130 at and 565/565°C entirely consist of perlitic steels. Titanium alloys (Table 1) are used for the blades of the last stage of the K-300-240 (K-300-240) turbine (length 866 mm). To make possible the use of perlitic steels also for 600-610°C, new heat-resistant steels were developed on the basis of 12% chrome steels. In order to reduce liquefaction phenomena, the TsNIITMASH is modifying УМ-5(TsZh-5) steel with calcium. Austenitic steels, alloyed nickel steels and perlitic steels with cooling are used for the СКР-100 (SKR-100) topping turbine for 100 Mw, 300 at, 650°C. At present, forgings of 3-4 tons are produced from ЭИ-612 (EI-612) steel, and such of up to 13 tons from ЭИ-726 (EI-726) steel. Owing to the high coefficients of linear expansion and low thermal conductivity of the austenitic steels, the starting of the SKR-100 turbine is problematic, and additional heating of the flange joints must be provided. For austenitic steels at temperatures of 650 and 750°C, the rate of oxidation of the turbine steels in air lies in the order of magnitude of thousandths and hundredths of millimeters annually. There are 7 figures and 3 tables.

Table 1 Mechanical properties of titanium alloys. Legend: (?) content
Card 2/8

S/096/61/000/002/009/014
E111/E194

AUTHORS: Lipshteyn, R.A., Candidate of Technical Sciences,
Khaykina, S.E., Candidate of Technical Sciences, and
Ginzburg, E.S., Candidate of Technical Sciences

TITLE: Vanadium Corrosion in Boiler Installations

PERIODICAL: Teploenergetika, 1961, No.2, pp. 61-62

TEXT: The authors show that vanadium corrosion of boiler tubes working on high sulphur fuel oils is appreciable. Results are shown in Table 1 and give comparative data on corrosion of type 381T(EYalT) steel in 60 hours at 750 °C by artificial and real deposits. Previous work (Ref.1) suggested that corrosion did not occur if there was no oxygen in the gases. The present investigation was undertaken to study the influence of oxygen concentration in the gas. Type 314-405 (EI-405) steel (0.11% C, 0.46% Si, 0.72% Mn, 14.1% Cr, 13.2% Ni, 1.36% Nb and 2.5% Mo) was used. The washed and dried 6 mm diameter, 30 mm long cylindrical specimen was weighed and then, while embedded in an artificial ash (87% V₂O₅, 13% Na₂SO₄) at 800 ± 5 °C, was subjected to the action of a nitrogen-oxygen mixture (up to about 95% O₂). ✓

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Vanadium Corrosion in Boiler Installations

The apparatus (figure, page 62) provides for measurement of gas-volume changes produced by reaction with the specimen/ash. Specimen weight changes were also determined. The results (Table 2) show that the higher the oxygen content the greater the corrosion. The results suggest that combustion gases with 3-4% oxygen will produce vanadium corrosion if metal surface temperatures are over 650 °C and the deposits are relatively high in vanadium. Corrosion will start on superheater and radiation tubes.

There are 1 figure, 2 tables and 3 references: 2 Soviet and 1 English.

ASSOCIATION: Vsesoyuznyy teplotekhnicheskiy institut
(All-Union Heat Engineering Institute)

Card 2/2

"APPROVED FOR RELEASE: Thursday, September 26, 2002

CIA-RDP86-00513R000515120015-1

APPROVED FOR RELEASE: Thursday, September 26, 2002

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CIA-RDP86-00513R000515120015-1
CIA-RDP86-00513R000515120015-1"

BOREVSKIY, Ye.I., inzh.; OSTROVSKIY, S.I., inzh.; GINSBURG, E.P., kand.
tekhn. nauk

Study of the performance of metal and the construction of the
gland when starting a R-100-300 KtGC steam turbine. Teplo-
energetika 10 no.10:13-18 0163
(MIRA 17:7)

1. Vsesoyuznyy otdeleni Trudovogo Kraesnog Znameni teplotekhnichesk-
kiy Institut imeni Dzerzhinskogo i Khar'kovskiy turbinnyy zavod
imeni S.M. Kirova.

BERMAN, L.D., doktor tekhn. nauk; GINZBURG, E.S., kand. tekhn. nauk;
DUBNITSKAYA, L.Ye., inzh.; PROKHOROVA, Ye.I., inzh.

Operational tests of tubes from aluminum alloys in condensers and
water heaters. Elek. sta. 34 no.5:28-32 My '63. (MIRA 16:7)

(Pipes, Aluminum—Corrosion)
(Condensers (Steam))

GINZBURG, E.G.

The use of carotene in therapy. F. O. Ginzburg.
Klin. Med. (U. S. S. R.) 16, 1342-6 (1937); *Chem. Zentr.*
1939, I, 2010.—After a review of the chem. and physiol.
properties of carotenoids, the analgesic, epithelializing, and
disinfecting action of the carotene-contg. preps. Vitaderm
and Caroton are pointed out. M. G. Moore

OPEN

INTERNAL INDEX

430-114 METALLURICAL LITERATURE CLASSIFICATION											
EIGHTH EDITION											
SEARCH SYSTEM											
SEARCH NO.	1	2	3	4	5	6	7	8	9	10	11
SEARCH NO.	W	D	D	W	R	R	R	R	R	R	R

SEARCH NUMBER

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SEARCH NUMBER

GINZBURG, E.G.

11F

ca

State of potassium in an erythrocyte. Yu. A. Kriger
and E. G. Glusberg (Hospital Inst. Hematology) Pro-
letarsky, Kiev, Moscow. Ryad. Akad. Med. Nauk.
18, No. 3/6, AS, 7 (1944). By the method of Borodanskaya
(Ryad. Akad. Med. Med. 17, No. 4/5, 1944) it was shown
that in human and rabbit blood in hypotonic hemolysis
K ions are liberated from erythrocytes before hemo-
globin. G. Lebedeff

APPENDIX - METALLURGICAL LITERATURE CLASSIFICATION

CLASSIFICATION	SUBDIVISION	SIGHT DENSITY	GENERAL SUBJECT INDEX																	
			A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R
GENERAL	GENERAL	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19

"APPROVED FOR RELEASE: Thursday, September 26, 2002
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CIA-RDP86-00513R000515120015-1
CIA-RDP86-00513R000515120015-1"

DERVIZ, G.V.; GINZBURG, F.G.

"Methods of chemical blood analysis." S.D. Balakhovskii, I.S.
Balakhovskii. Reviewed by G.V. Derviz, F.G. Ginzburg. Biokhimiia
20 no.6:749-752 N-D '55.
(MIRA 9:3)

(BLOOD--ANALYSIS AND CHEMISTRY) (BALAKHOVSKII, S.D.)
(BALAKHOVSKII, I.S.)

*GINZBURG, F.G.*USSR/Human and Animal Physiology - Blood. Blood Transfusions
and Blood Substitutes.

T-4

Abs Jour : Ref Zhur - Biol., No 10, 1958, 45943

Author : Ginzburg, F.G.Inst :
Title : Specific Permeability of Erythrocytes and Its Significance for Their Preservation.Orig Pub : V sb.: Sovrem. probl. gematol. i perelivaniya krovi.
Vyp. 32, M., Medgiz, 1956, 62-68.

Abstract : As a 5 percent glucose solution (I) was added to human erythrocytes (E) at a temperature of 6°C, they became enlarged by 37-49 percent (about 43 percent on the average) in their size, while at a temperature of 18° C, they hemolyzed rapidly because of permeability by I. At a temperature of 6° C, E size did not change perceptibly in goats, and at a temperature of 18° C it became only slightly enlarged after 24 hours, a phenomenon

Card 1/2

VINOGRAD-PINKHLI, F.R., prof.; QINZBURG, F.O.; FEODOROVA, L.I.; KAUKHCHISHVILI,
L.I.

Blood preservation at temperatures lower than 0° C; preliminary
report [with summary in English, p.61-62] Probl.gemat. i perel.
krovi 3 no.1:27-34 Ja-F '58. (MIRA 11:3)

1. Iz TSentral'nogo ordena Lenina instituta hematologii i pereliva-
niya krovi (dir. - deyatel'nyy chlen AMN SSSR prof. A.A.Bagda-
sarov) Ministerstva zdravookhraneniya SSSR.

(BLOOD PRESERVED,
eff. of cold (Rus))

VINOGRAD-FINKEL', F.R., prof.; GINZBURG, F.G.; FRDOROVA, L.I.

Preservation of blood in frozen state. Akt.vop.perel.krovi no.7:
91-97 '59. (MIRA 13:1)

1. Laboratoriya konservirovaniya krovi (zav. laboratoriye - prof.
F.R. Vinograd-Finkel') i biokhimicheskaya laboratoriya (zav. labora-
toriye - prof. G.V. Derviz) TSentral'nogo instituta hematologii i
perelivaniya krovi.
(BLOOD--COLLECTION AND PRESERVATION)

VINOGRAD-FINKEL', F.R., professor, kand.biologicheskikh nauk;
GINZBURG, F.G.; FEDOROVA, L.I.; KAUKHCHESHVILI, E.I.

Low-temperature preservation of blood. Priroda 49
no.7:88-89 Jl '60. (MIRA 13:7)

1. TSentral'nyy institut hematologii i perelivaniya krovi,
Moskva (for Fedorova). 2. Moskovskiy tekhnologicheskiy
institut myasnoy i molochnoy promyshlennosti (for Kaukhchesh-
vili).
(BLOOD--COLLECTION AND PRESERVATION)

VINOGRAD-FINKEL', F.K., prof.; KISELEV, A.Ye., dotsent; CHINZBURG, P.G.;
FEDOROVA, L.I.; KAUKHCHUSHVILLI, E.I.

Use of deepfreeze for the prolonged preservation of blood in
a frozen state. Probl. gemat. i perel. krovi 8 no.5:3-16
(MIRA 16:8)
May'63.

1. Iz TSentral'nogo ordena Lenina instituta hematologii i
perelivaniya krovi (direktor - dotsent A.Ye.Kiselev) Mini-
sterstva zdravookhraneniya SSSR.
(BLOOD—COLLECTION AND PRESERVATION)

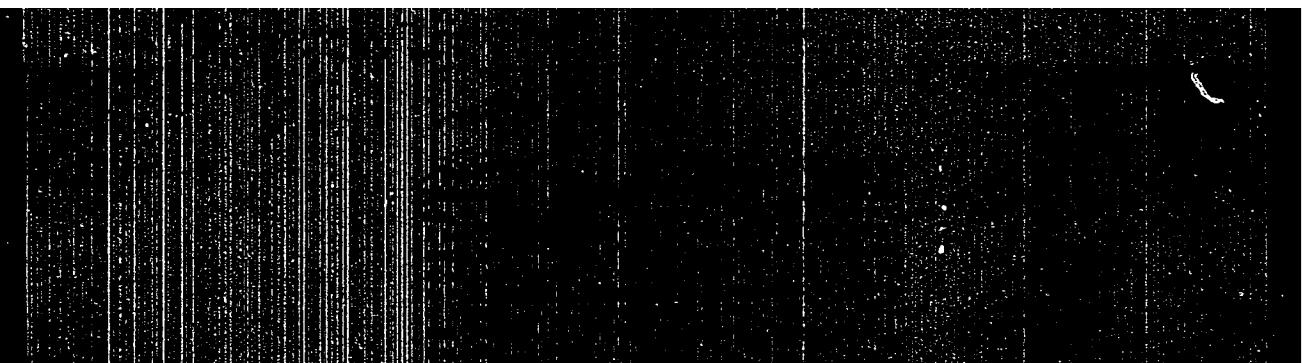
VINOGRAD-FINKEL', F.R., prof.; KISELEV, A. Ye. docent. GINZBURG, F.G.,
FEDOROVA, L.I.; SEMENOVA, N.V.; KOMALEK, E.I.; BURDYAGA, F.A.
TALESKAYA, I.N.; KUDRYAVCEVA, S.N.

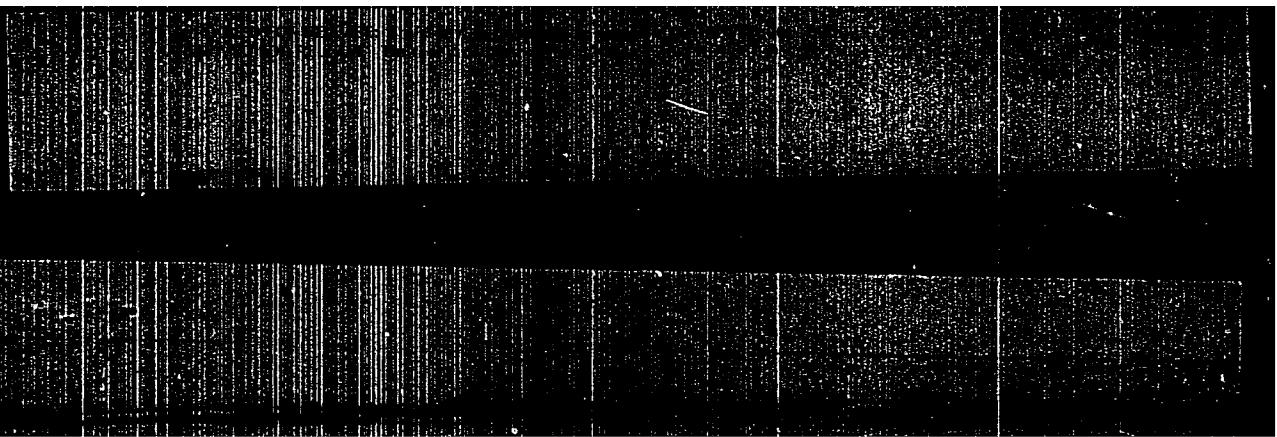
Long-term preservation of blood in frozen state. Vopr.-med. zhur.
(VOPR. MED. ZHUR.)
no. 1:27-33 Ja 1966

Ma.
Review of Theory

On the Theory of Superconductivity. V. L. Ginsberg and L. D. Landau (Zhur. Khim. Fiz., 1930, 26, 112; 1934-1935).--[In Russian]. The existing phenomenological theory of superconductivity is unsatisfactory, because it does not enable the surface tension at the boundary between the normal and superconductive phases to be determined, and because it cannot give a correct description of the destruction of superconductivity by a magnetic field or by an electric current. G. and L. attempt to construct a theory free from these defects. Equations are deduced for the "Y" functions of the "superconductive electrons" introduced into the theory, and for the vector potential. A soln. of the equations is given for the one-dimensional case (a semi-infinite supra-

conductive region or a superconductive lamina). The theory enables the surface tension to be expressed in terms of the critical magnetic field and the depth of penetration of a magnetic field into the superconductor. In a strong field, the depth of penetration depends on the strength of the field, and this effect should be clearly discernible in superconductors of small dimensions. The destruction of superconductivity in thin lamina by a magnetic field proceeds by means of a 2nd-order phase transition, while only in laminae of thickness greater than a certain critical value is the transition of the 1st order. Though the critical external magnetic field increases with decreasing thickness of the lamina, the critical current, which destroys superconductivity in the lamina, decreases with decreasing thickness.--O. B. H.





"APPROVED FOR RELEASE: Thursday, September 26, 2002

CIA-RDP86-00513R000515120015-1

APPROVED FOR RELEASE: Thursday, September 26, 2002

CIA-RDP86-00513R000515120015-1"

STARIK, I.Ye.; GINZBURG, F.L.

State of microquantities of radioelements in dilute solutions.
Part B: Adsorption of lanthanum on quartz glass and plexiglas.
Radiokhimia 1 no.2:171-173 '59. (MIRA 12:8)
(Lanthanum) (Adsorption)

STARIK, I.Ye.; GINZBURG, F.L.

State of microquantities of radioelements in solutions. Part 14:
Study of the state of americium in aqueous solutions. Radiokhimia
1 no.4:435-438 '59.
(Americium)

APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515120015-1
APPROVED FOR RELEASE: Thursday, September 26, 2002, CIA-RDP86-00513R000515120015-1"
STARIK, I.Ye.; RATNER, A.P. [deceased]; PASVIK, M.A. [deceased]; GINZBURG, F.L.

Use of phenylarsonic acid for the separation of neptunium and
plutonium. Radiokhimia 1 no.5:545-547 '59. (MIRA 13:2)
(Benzencarsonic acid) (Neptunium) (Plutonium)

21(0)

AUTHORS:

Ginzburg, F. L., Rozovskaya, N. G.

SOV/30-59-6-29/40

TITLE:

The State of Microquantities of Radioelements in Solutions
(Sostoyaniye mikrokolichestv radioelementov v rastvorakh)

PERIODICAL:

Vestnik Akademii nauk SSSR, 1959, Nr 6, pp 122-124 (USSR)

ABSTRACT:

These problems were the subject of an All-Union Symposium held in Leningrad from March 3 to March 5. I. Ye. Starik spoke about the lack of interest in the research of the molecular form of elements. I. Ye. Starik, I. A. Skul'skiy, N. I. Ampelogova, L. I. Il'men'kova, L. D. Sheydina and F. L. Ginzburg reported on the investigation of the state of the microquantities of zirconium, polonium, protactinium and americium in aqueous solutions. M. N. Yakovleva and M. A. Shushalina delivered reports on the methods of investigating the state of uranium in natural waters. V. M. Vdovenko, L. N. Lazarev and S. Ya. Khvorostin dealt in their report with the investigation of the state of radioelements in nonaqueous phases. V. M. Vdovenko, Ye. A. Smirnova and N. A. Alekseyeva spoke about the degree of hydration of complex compounds of uranyl nitrite and nitric acid in organic solvents. A new method of determining the composition of complex compounds and the calculations of the instability ✓

Card 1/2

The State of Microquantities of Radioelements in Solutions SOV/30-59-6-29/40

constants was recommended by V. M. Vdovenko, A. A. Chaykhorskiy and L. M. Belov. A. K. Lavrukhina showed that the forms of existence of a radioelement depend on its concentration in the solution. V. I. Kuznetsov and P. D. Titov explained the effect of the co-extracting by the formation of mixed polyanions. A. M. Trofimov and L. N. Stepanova recommended a method of determining the degree of ionic charge of radioelements in a solution. S. Ye. Bresler, Yu. D. Sinochkin, A. I. Yegorov and D. A. Perumov showed that the use of specific sorbents on zirconium basis may be of practical value for the investigation of the form of radioelements in solutions. An. N. Nesmeyanov dealt with the substitution of hydrogen in benzene by the atoms P^{32} , As^{76} , Sb^{124} . V. M. Vdovenko emphasized the great interest displayed by the scientific public in this Symposium in the name of the Organization Committee and said that approximately 250 scientific collaborators contributed to the work carried out by it.

✓

Card 2/2

"APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515120015-1
APPROVED FOR RELEASE: Thursday, September 26, 2002 CIA-RDP86-00513R000515120015-1"

STARIK, I.Ye.; AMPELOGOVA, N.I.; GINZBURG, F.L.; LAMBERT, M.S.; SKUL'SKIY, I.A.;
SHCHERBETKOVSKIY, V.N.

Molecular state of ultraminiute quantities of radioelements in
solutions. Radiokhimia 1 no.4:370-378 '59. (MIRA 13:1)
(Radioactive substances)

GINZBURG, F.L.; ROZOVSAYA, N.G.

State of microscopic quantities of radioelements in solutions.
Vest. AN SSSR 29 no.6:122-124 Je '59. (MIRA 12:5)
(Radioisotopes) (Solutions (Chemistry))

23876

3/185/61/004/001/009/010
AUG 1 A129

213230

AUTHORS: Starik, I.Ya., Ginzburg, F.L.

TITLE: The state of microquantities of radioelement in diluted solutions
XVI. An investigation of the state of americium by the ion-exchange
method

PERIODICAL: Radikal'nost', v 3, no 1, 1961, 45-50

TEXT. The authors conducted a detailed study on the behavior of americium when using ion-exchanging resins and compared the obtained data with previously derived theories of americium behavior in diluted solutions (Ref 10-12). The main considerations were given to the cation exchange from the concentrated HCl as a washing-out solution has a great effect in group separation of actinides and rare-earth elements adsorbed on the radionuclides. The authors used the method of ion-exchange for studying the state of americium in nitrate solutions. The sorption of Am²⁴¹ on resins under static condi-

Card 1/1

S/186/61/001/009/C20
A051/A129

The state of microquantities...

Am was investigated depending on the pH of the solution and concentration of HNO_3 . The KU-2 (EU-2) and Dowex-50 cationites were used as the adsorbents, as well as ionites of the strong-acid type with a basic-functional sulfon-group, AB-17 (AV-17) and Dowex-1 anionites containing strongly dissociated active amino groups of the tetra-aminium base. The absorption of Am from solutions of various pH was conducted on KU-2 resin in the potassium form and AV-17 in the NO_3^- form. The sorbability of Am in the ionites was determined at room temperature under conditions of achieving an adsorption equilibrium state. The concentration of Am in the water phase was determined according to the activity of an aliquot part of the solution. The results of the experiments were expressed in % of adsorption determined by the formula:

percentage of adsorption = $\frac{A_2 - A_1}{A_2} \cdot 100$, where A_2 is the initial activity of the solution (in pulses/min), A_1 - the equilibrium activity of the solution (in pulses/min). The relationship of Am sorption to the pH of solution was studied on the KU-2 cationite and AV-17 anionite. The sorption was conducted from solutions in the presence of 10^{-3}M KNO_3 . Fig 1 shows the relationship of Am sorption on KU-2 resin to the pH of the solution. It is seen

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The state of nitrogen fixation is

that at pH 7.0 it may stop in the transition state because of an increase in the reactivity of the molecule. At pH 7.0 it was observed that the formation of a hydrogen bond between the NH₂ group and the carbonyl oxygen was complete. It was also observed that the formation process of a hydrogen bond began at pH 1.54. At pH 7.0 the hydrogen bond was almost completely formed. The deactivation of the hydride donor by the NH₂ group decreased the rate of reduction by the hydride acceptor. The rate of reduction increased when the NH₂ group was removed from the hydride donor. The reaction mechanism of the reduction of the hydride donor by the hydride acceptor was studied by the authors. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor. The authors also studied the effect of the hydride acceptor on the rate of reduction of the hydride donor.

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The state of equilibrium is

by the last condition, the total solubility of HNO₃, taken into account, initial dehydratation of water and the nature of salts present in solution. It is thought that the concentration of HNO₃ in the solution may decrease. Applying the law of Raoult's we obtain the following relations between the ion concentrations of a salt and the equilibrium constant of resin partition function, representing balance of ion exchange reaction, the following can easily be derived:

It is known that the equilibrium constant of the ion exchange reaction is proportional to the square of the ion charge of the resin, changing the value of the slope of the straight line, expressed by the equation

$\log K = \log K_0 + \frac{1}{2} \log \left(\frac{c_{\text{HNO}_3}}{c_{\text{HNO}_3}^0} \right)$

The ion charge directed by the resin from 0.5 to 0.55, which is equal to 3. In addition to complex I is there are always some ions present in the solutions and the dehydrated charge of the anion resin in the resin phase is determined by the shift of equilibrium according to equation of the actions by the resins. The authors draw the following conclusions:

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The state of microquantities ...

the sorption of Am²⁴¹ from nitrate solutions (1-16 M) in solutions of various pH, it is shown that in solutions of pH=4 to 1 M HNO₃, simple Am ions prevail with a charge of 3⁺; with an increase of the pH of the solution (pH > 4) the positive charge of the ions decreases due to hydrolysis; 2) in the colloidal state Am is not adsorbed by the cation; the sorption of the colloidal particles of Am by the anionite reaches 80%; 3) the results of investigations of electromigration and sorption of Am on the anionites in nitrate solutions show that positively charged Am complexes are formed in solutions of 1-4 M HNO₃. Neutral complexes of Am are dominant in solutions of 4-16 M HNO₃. There are 6 figures, 1 table and 24 references: 12 Soviet-bloc, 12 non-Soviet-bloc.

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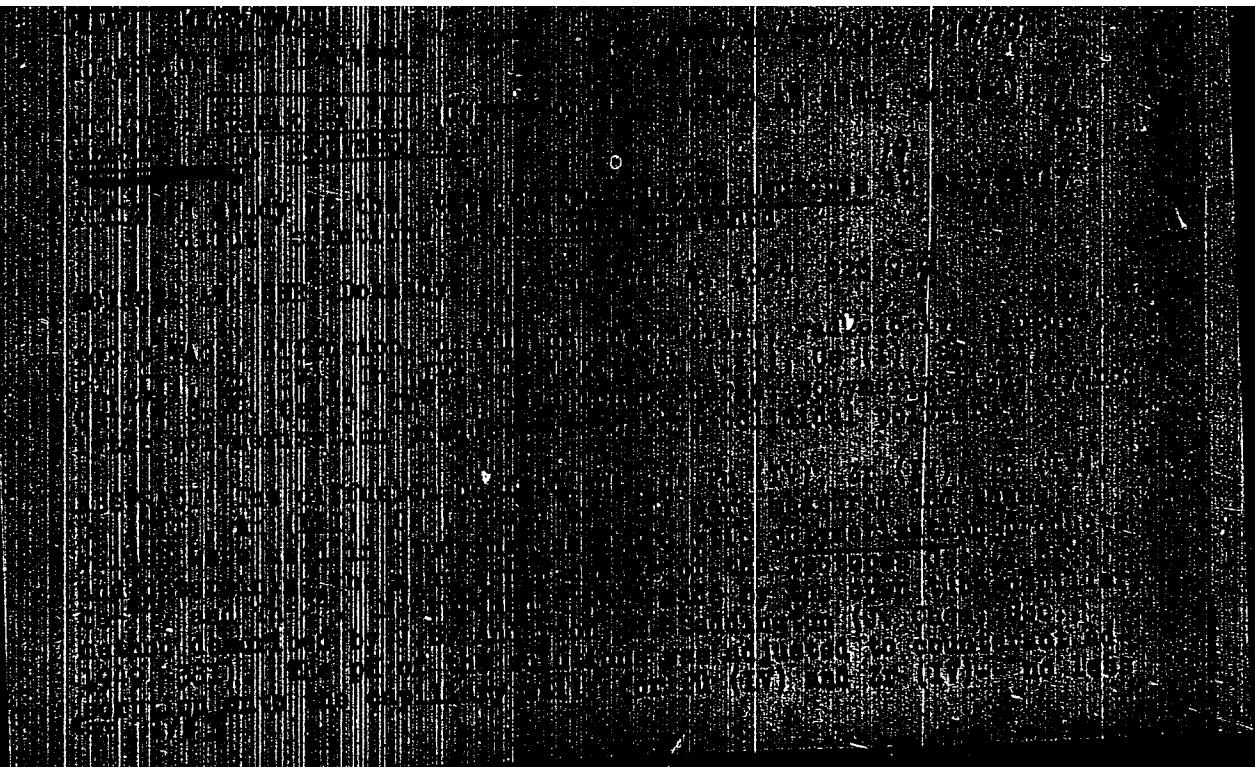
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